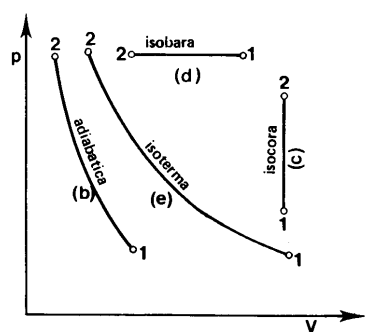


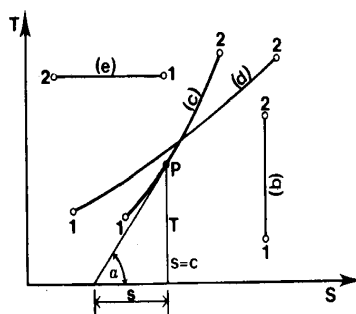
La politropica è una trasformazione a calore specifico costante per gas ideali. Vediamo dei casi particolari:

$m = 1$	$T = \text{cost}$	isoterma
$m = 0$	$P = \text{cost}$	isobara
$m = k$		isoentropica
$m = \infty$		isocora

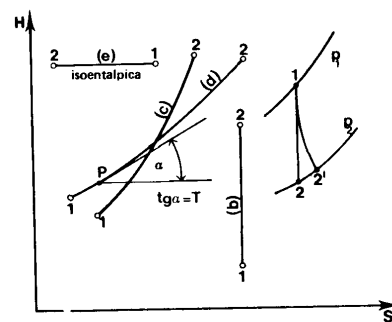
Rappresentazione delle trasformazioni nei piani termodinamici:



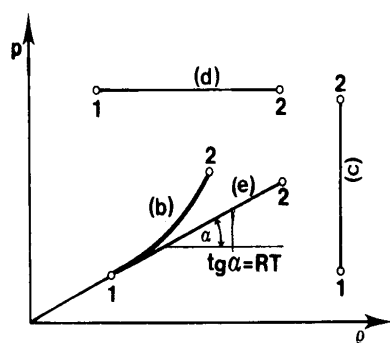
PV



TS



HS



Pp

b = adiabat
c = isocora
d = isobara
e = isoterma